Measurement of a charged particle

How to detect a charged particle

- Please remember "Passage of Radiation through Matter"!
 - The Bethe-Bloch Formula

$$-\frac{dE}{dx} = 2\pi N_{\rm a} r_{\rm e}^2 m_{\rm e} c^2 \rho \frac{Z}{A} \frac{z^2}{\beta^2} \left[\ln \left(\frac{2m_{\rm e} \gamma^2 v^2 W_{\rm max}}{I^2} \right) - 2\beta^2 - \delta - 2\frac{C}{Z} \right]$$

The charged particle interacts with an electron in an atom.
 The atom will be excited or ionized



How to detect a charged particle

- Observe
 - the free electron after ionization (or cation)
 - the excited atom



(1) Detection of a free electron (Cation) after lonization

1.Gaseous Ionization detector with noble gas.

- 1. Mean energy for ion-electron pair creation (W value)
 - ~26 eV for Ar (typically ~30 eV)
- 2. Catch the electric signal when the electrons move to an electrode
- 3. The signal of one electron is too small to be detected
 - 1. Need an amplification
- 4. The amplification depends on the strength of electric field

	Excitation potential	Ionization potential	Mean energy for ion-electron pair creation
	[eV]	[eV]	[eV]
H_2	10.8	15.4	37
He	19.8	24.6	41
N_2	8.1	15.5	35
$\tilde{O_2}$	7.9	12.2	31
Nē	16.6	21.6	36
Ar	11.6	15.8	26
Kr	10.0	14.0	24
Xe	8.4	12.1	22
CO_2	10.0	13.7	33
CH₄		13.1	28
$C_4 H_{10}$		10.8	23

Table 6.1. Excitation and ionization characteristics of various gases



(1) Detection of a free electron (Cation) after lonization

[Q1] How much is the energy deposit in 5mm thick gas?

[A1] $E \sim 2MeV/cm \times 0.5[cm] \times 0.001 [g/cm^2] = 1 keV$

[Q2] How many ion-electron pairs are created? [A2] E/W = $1000/26 \sim 40$ paris

[Q3] How much current flows in the circuit? Cosmic ray (MIP) +1kVThe electron drift velocity is assumed to be $2cm/\mu s$ $5mm \int e^{i}$ Ar:CO₂=90:10

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(1) Detection of a free electron (Cation) after Ionization[A3]

(1) The electric field moves an electron and an ion to the electrode.(2) The current flows to the condenser to keep the electric field.

ev

d

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• For a single ion-electron pair,

$$-eE \times dx = VdQ$$
$$I = dQ/dt = -eEv/V = -$$

I=0.6pA continues for the maximum period of 250ns.

If 40 ion-electron pairs are created,
 I=24pA (24×10⁻¹²A) is drawn.



(1) Detection of a free electron (Cation) after Ionization

[Remember] The source of current flow is the movement of electric charge!

- [Q] What moves the electric charges?
 - [A] The energy stored in the detector provided by the High-Voltage module.

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Amplification \rightarrow Avalanche Multiplication

Amplification \rightarrow Avalanche formation



Multiplication in gas

- \sim 1 : Ionization chamber (no multiplication)
- $10^3 \sim 10^5$: Proportional counter (The signal is proportional to the number of ion-electron pairs created in the beginning.)
- >10⁷: Geiger-Muller counter (The signal size plateaus because of discharge caused by avalanches spread out over all area of the detector. It can be used to count the number of incident particles.)





Single Wire Proportional Counter Cylindrical Proportional Counter



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by Masaya Ishino (ICEPP, U. of Tokyo)

Cylindrical Proportional Counter

Thin wire \Rightarrow High electric field around the wire

(exercise of Electromagnetism)

$$\mathbf{E} = \frac{Q/L}{2\pi\epsilon_0} \cdot \frac{1}{r}$$

$$V_0 = \int_{r_i}^{r_0} \mathbf{E}_{\Delta} dr = \frac{Q/L}{2\pi\epsilon_0} \cdot \frac{ln(r_0 r_2)}{r_1} \mathbf{E}_{C} dr = \frac{Q/L}{ln(\frac{r_0}{r_i})}$$



When an electron moves from r_1 to r_2 , the energy obtained from the electric field r_3

When the energy obtained is bigger than W_i in a short distance than the mean free path. The avalanche multiplication starts

by Masaya Ishino (ICEPP, U. of Tokyo)

Avalanche Multiplication portional Counter



When the electron reaches near (~ 300μ m) the wire, the avalanche starts

⁶⁵ by Masaya Ishino (ICEPP, U. of Tokyo)

Time evolution of the avalanche



- (a): the electron is forced to the wire
 - (b): Near the wire, E is large enough to generate avalanche
 - (c): μ _{電子} >> μ _{イオン} ⇒ Electrons and lons are separated
 - (d) : Surrounding the wire with water-drop shape (Inner: electrons, Outer: ions)
 - (e) : Electrons are absorbed i the wire within 1ns, while ions are drifting to the wall with OV.

by Masaya Ishino (ICEPP, U. of Tokyo)



- 直径 0.25 mm の注射針である (Bennett and Yule⁴⁾ による)
- Typical gas mixture: 90% Ar and 10% CH₄
 - Avoiding the electron captured in the gas (Nobel gas, such as Ar).
 - Oxygen is not good
 - Quencher (such as CH₄) is necessary
 - When the atom is excited, it generates photons which ionize another atom. If the process continues, too much avalanches are produced. To avoid this process, quenching gas is used to absorb photons.
- The selection of gas is not straightforward by considering the drift velocity of electrons as a function of voltage, multiplication, amount of material, possibility of discharge, etc.. You must be an expert.

Geiger-Muller counter



- Typical gas: Ar
- with quencher
 - the function of quencher is not same as that used in a promotional counter







(2) Observation of excited atoms

- Detect photons emitted in deexcitation process of excited atoms (or molecules).
 - Fluorescent light
- \cdot The photon produced by radiation is called as a scintillation photon. The material that emits the scintillation photons is called as scintillator.
- \cdot Characteristics of good scintillators are
 - High efficiency to transform the energy deposit by radiation to scintillation photons (even so, 10% is high)
 - The amount of scintillation photons is proportional to the original energy deposit (linear relation).
 - The material is transparent for scintillation photons.
 - The scintillation photons are produced within a short time period of nsec~ μ sec (The life time of excited atoms is short).

Type of scintillators

- \cdot Organic Scintillator
 - •The scintillation photons are emitted from transitions made by the free valence electrons of the molecules. Scintillation material can be in the states of solid, liquid and gas.
- Inorganic Crystal scintillator
 - The scintillation photons are emitted based on the electron band structure of inorganic crystals.







Fig. 7.4. Energy level diagram of an organic scintillator molecule. For clarity, the singlet states (denoted by S) are separated from the triplet states (denoted by T)

Inorganic Crystal Scintillator



Fig. 7.7. Electronic band structure of inorganic crystals. Besides the formation of free electrons and holes, loosely coupled electron-hole pairs known as excitons are formed. Excitons can migrate through the crystal and be captured by impurity centers

- \cdot Organic scintillator
 - The organic scintillators are aromatic hydrocarbon compounds containing linked or condensed benzene-ring structures.



- Scintillation lights are little absorbed (transparent)
- The emission time of scintillation lights (the lifetime of excited states) is fast [2~3 nsec].



Scintillation efficiency

 The efficiency is getting low if the excited energy is not used for the emission of lights, but heat. The process is called as quenching.
 (Example) Quenching occurs in the liquid scintillator with Oxygen.

Solvent and Solute

- •The ionization energy seems to be absorbed mainly by the solvent (and plastics) and then passed on to the scintillation solute. This transfer is quick and efficient. A typical scintillation solute is PBD, PPO and POPOP.
- \cdot Wave Length Shifter
 - •The secondary solute such as POPOP is added with the first solute of PBD for its wavelength shifting properties. The primary scintillation photons are absorbed by the secondary solute and emits the photons of longer wavelength that are more transparent and more matched to the sensitivity of photon-sensors.

• Organic Crystal Scintillator: Anthracene (C14H10), etc..

Plastic Scintillator

• Very Flexible shape: Scintillator plate, Scintillator bar (T2K ND280/FGD), Scintillating Fibers (NINJA Tracker), etc..

- Organic Liquid Scintillator
 - ·KamLAND, Double Chooz, Daya Bay, NOvA, etc..





	Scintillator	Туре	Density	Re- frac- tive index	Melting softening or boiling point C ^a	Light output (% Anthra- cene)	Decay constant, main com- ponent [ns]	Wave- length of maxi- mum emission [nm]	Content of loading element (% by wt.)	H/C No. of H atoms/ No. of C atoms	Principal applications
Plastic	NE 102A	Plastic	1.032	1.581	75	65	2.4	423		1.104	ν , a , β , fast n
	NE 104	Plastic	1.032	1.581	75	68	1.9	406		1.100	ultra-fast counting
	NE 104B	Plastic	1.032	1.58	75	59	3.0	406		1.107	with BBO light guides
	NE 105	Plastic	1.037	1.58	75	46		423		1.098	dosimetry
	NE 110	Plastic	1.032	1.58	75	60	3.3	434		1.104	γ , α , β , fast n etc.
	NE 111A	Plastic	1.032	1.58	75	55	1.6	370		1.103	ultra-fast timing
	NE 114	Plastic	1.032	1.58	75	50	4.0	434		1.109	as for NE 110
	NE 160	Plastic	1.032	1.58	80	59	2.3	423		1.105	use at high temperatures
	Pilot U	Plastic	1.032	1.58	75	67	1.36	39 1		1.100	ultra fast timing
	Pilot 425	Plastic	1.19	1.49	100			425		1.6	Cherenkov detector
Liquid	NE 213	Liquid	0.874	1.508	141	78	3.7	425		1.213	fast n (P.S.D.)
	NE 216	Liquid	0.885	1.523	141	78	3.5	425		1.171	a, β (internal counting)
	NE 220	Liquid	1.036	1.442	104	65	3.8	425	O 29%	1.669	internal counting, dosimetry
	NE 221	Gel	1.08	1.442	104	55	4	425		1.669	α , β (internal counting)
	NE 224	Liquid	0.877	1.505	169	80	2.6	425		1.330	y, fast n
	NE 226	Liquid	1.61	1.38	80	20	3.3	430		0	y, insensitive to n
	NE 228	Liquid	0.71	1.403	99	45		385		2.11	n
	NE 230	Deuterated liquid	0.945	1.50	81	60	3.0	425	D 14.2%	0.984	(D/C) special applications
	NE 232	Deuterated liquid	0.89	1.43	81	60	4	430	D 24.5%	1 .96	(D/C) special applications
	NE 233	Liquid	0.874	1.506	117	74	3.7	425		1.118	α , β (internal counting)
	NE 235	Liquid	0.858	1.47	350	40	4	420		2.0	large tanks
	NE 250	Liquid	1.035	1.452	104	50	4	425	O 32%	1.760	internal counting, dosimetry
Loaded	NE 311 & 311A	B loaded liquid	0.91	1.411	85	65	3.8	425	B 5%	1.701	п. в
liquid	NE 313	Gd loaded liquid	0.88	1.506	136	62	4.0	425	Gd 0.5%	1.220	n
	NE 316	Sn loaded liquid	0.93	1.496	148.5	35	4.0	425	Sn 10%	1.411	v. x-ravs
	NE 323	Gd loaded liquid	0.879	1.50	161	60	3.8	425	Gd 0.5%	1.377	n

Table 7.1. Physical properties of various commercial scintillators (data from Nuclear Enterprises scintillator catalog [7.1])

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Response of organic scintillator

\cdot Time response

- · Fast (2~3nsec)
- The timing shape sometimes depends on the energy deposit (dE/dx). This character is used for Particle Identification. (neutron/ γ with liquid scintillator, proton/ μ , π in T2K ND280&INGRID)
- · Energy Response
 - $\cdot\,A$ dependence of the energy deposit (dE/dx) by the quenching effect.



Fig. 7.3. Resolving scintillation light into *fast* (promp (delayed) components. The *solid line* represents the total curve

σ [ns]	τ [ns]
0.7	2.4
0.2	1.7
0.5	1.87
	σ [ns] 0.7 0.2 0.5

Table 7.2. Gaussian and exponentialparameters for light pulse descriptionfrom severalplasticscintillators(from Bengston and Moszynski [7.2])

$$\frac{I}{I_0} = f(t) \cdot e^{-t/t}$$

f(t): Gaussían wíth σ ет



(3) Detection of a free electron after Excitation

- 1. A semiconductor detector is very popular in High energy physics experiments.
 - 1. Measure the electric signal originated by electron-hole pairs created by energy deposit of a charged particle.
 - 2. The energy need for an electron-hole pair is small and we can measure many electron-hole pairs.
 - Energy for electron-hole pair creation in semiconductor: ~3 eV
 - \cdot Energy for electron-ion pair creation in gas: ~30 eV
 - \cdot Energy for scintillation photon : ~100 eV



Fig. 10.1. Energy band structure of conductors, insulators and semiconductors

What shall we measure for the charged particle

With the signal of the charged particle, we can measure

- 1. position
- 2. momentum
- 3. velocity
- 4. energy

of the particle.

Measurement of a neutral particle (mainly a photon)

How to detect a neutral particle, photon

- Remember the interaction of photons with matters
 - (1) Photoelectric Effect [in low energy]
 - (2) Compton Scattering [in medium energy]
 - (3) Pair production [in high energy]
 - In the interaction, an electron is emitted. We will measure the electron as a signal of photon interaction. We already know how to detect a charged particle.



γ ray spectrum measurement

- Signal (Ideal situation)
 - Photoelectric peak
 - Compton spectrum with edge
 - Escape peak of positronium annihilation
 - Single escape peak
 - Double escape peak





γ ray spectrum measurement

- Signal (Ideal situation)
 - Photoelectric peak
 - Compton spectrum with edge
 - Escape peak of positronium annihilation
 - Single escape peak
 - Double escape peak



How to detect a photon

- First, a photon is interacted, and second, the electron is measured.
- Inorganic crystal scintillator
 - The advantage lies in the greater stopping power due to the higher density and higher atomic number. [Q] Why the higher atomic number?

Atomic number		den sity	wave- length (nm)	reflection index	decay time (µs)	light yield
• Cd: 48	アルカリハライド			Andreich an Antonio an Anna an		
• l· 53	NaI (Tl)	3.67	415	1.85	0.23	38,000
• Cs: 55	CsI (Tl)	4.51	540	1.80	0.68 (64 %) 3.34 (36 %)	65,000
$\sim 1 \wedge l \cdot 7 / l$	CsI (Na)	4.51	420	1.84	0.46, 4.18	39,000
• VV. 74	LiI (Eu)	4.08	470	1.96	1.4	11,000
• Bi: 83	その他の遅い無機シンラ	Fレータ	•			
	BGO	7.13	480	2.15	0.30	8,200
	$CdWO_4$	7.90	470	2.3	1.1 (40 %) 14.5 (60 %)	15,000

表 8.3 よく用いられる無機シンチレータの特性.

Scintillation mechanism

- Electronic band structure in crystals: [Example: Nal(TI)]
 - If a charged particle goes through the crystal, electrons are excited to conduction band from the valence band and the holes are also paircreated. An electron and hole pair is loosely coupled and forms an exciton.
 - It is not efficient that the electrons go back to the valence band with scintillation light. The probability is not high for an electron to meet a hole.
 - By doping impurities, the excitons can migrate through the crystal and be captured by impurity centers. Then, the scintillation lights are emitted.



Fig. 7.7. Electronic band structure of inorganic crystals. Besides the formation of free electrons and holes, loosely coupled electron-hole pairs known as excitons are formed. Excitons can migrate through the crystal and be captured by impurity centers

Scintillation mechanism

- Lifetime of the excitation state (typically 30~500 nsec) is longer than the time when the excitons are captured by impurity centers.
- The wavelength of scintillation lights is in a wider range. We often use the scintillator with visible lights.





Scintillation mechanism

- Light Yield is important. [Example: Nal(TI)]
 - In the case of 1MeV energy deposit
 - Efficiency of scintillation is typically 12%.
 - The energy of visible light is 3eV.
 - The number of scintillation lights is
 - 1,000,000eV×0.12%÷3eV=40,000
 - Since 1,000,000eV÷40,000 photons=25eV/photon, the energy for one electron-hole pair is 25eV.



The energy resolution depends on the light yield

many kinds of inorganic scintillators

		表 8.3 。	F<用いられ	る無徳シンチレー	タの特性		
	比重	最高放出 波長 (nm)	屈折率	减衰時間 (μs)	絶対発光量 (光子/MeV)	バイアルカリ 光電子増倍管 による相対的 パルス波高	参考文献
アルカリハライド							
NaI (TI)	3.67	415	1.85	0.23	38,000	1.00	
CsI (Tl)	4.51	540	1.80	0.68 (64 %) 3.34 (36 %)	65,000	0.49	85, 105, 106
CsI (Na)	4.51	420	1.84	0.46, 4.18	39,000	1.10	112
Lil (Eu)	4.08	470	1.96	1.4	11,000	0.23	
その他の遅い無機シンラ	チレータ						
BGO	7.13	480	2.15	0.30	8,200	0.13	
CdWO4	7.90	470	2.3	$\begin{array}{c} 1.1 \ (40 \ \%) \\ 14.5 \ (60 \ \%) \end{array}$	15,000	0.4	119-121
CaWO ₄	6.1	420	1.94	8	15,000		123
SrI ₂ (Eu)	4.6	435		1.2	85,000		125
ZnS (Ag) (多結晶)	4.09	450	2.36	0.2		1.3 ^a	
CaF ₂ (Eu)	3.19	435	1.47	6.0	24,000	0.5	
活性化物質なしの高速	無機シン	チレータ					
BaF ₂ (高速成分)	4.89	220		0.0006	1.400	na	133-135
$BaF_2 (遲 発 成 分)$	4.89	310	1.56	0.63	9,500	0.2	133-135
CsI (高速成分)	4.51	305		0.002 (35 %) 0.02 (65 %)	2,000	0.05	140142
CsI (遅発成分)	4.51	450	1.80	数/usまで多くの 成分	金で	金で	141, 142
CeF ₃	6.16	310, 340	1.68	0.005, 0.027	4,400	$0.04 \sim 0.05$	83, 146, 147
Ce 活性化高速無機シ	レチレータ						
GSO	6.71	440	1.85	$\begin{array}{c} 0.056\ (90\ \%)\\ 0.4\ (10\ \%)\end{array}$	6,000	0.2	156-160
YAP	5.37	370	1.95	0.027	18,000	0.45	85, 165
YAG	4.56	550	1.82	0.088 (72 %) 0.302 (28 %)	17,000	0.5	85, 167
LSO	7.4	420	1.82	0.047	25,000	0.75	170, 171
YSO	4.54	420		0.070	24,000		152, 153, 155
LuAP	8.4	365	1.94	0.017	17.000	0.3	178, 180, 183
LaCl ₃ (Ce)	3.79	350		0.028	46,000		212
LaBr ₃ (Ce)	5.29	380	$2.05 \sim 2.10$	0.026	63,000		212, 218
ガラスシンチレータ							
Ce 活性化 Li ガラス b	2.64	400	1.59	$0.05 \sim 0.1$	3,500	60:0	84, 241
Tb 活性化ガラス ^b	3.03	550	1.5	約3,000~5,000	約 50,000	na	241
比較用, 典型的な有機	(7°5,7,5	-ック) シンチ1	4-1				
NE102A	1.03	423	1.58	0.002	10,000	0.25	

配合により特性は変化,表15.1参照. 記するもの以外は文献⁸¹⁾⁸²⁾より主に引用.
Scintillation mechanism

- Temperature dependence (some energy is transformed to heat)
 - Light yield (Scintillation efficiency) varies.
 - Timing shape (decay time, lifetime of excited state) varies.



Photomultiplier Tube (PMT)

- An equipment to measure photons
- It often called as PMT (PhotoMulTiplier tube)

https://youtu.be/9EbX0dfWuU4





高エネルギー物理学分野への進出 - Foray into High Energy Physics Hamamatsu: No. 1 PMT vendor Prof. Koshiba with 50 cm PMT 「カミオカンデからスーパーカミオカンデへ」 戸塚 洋二 より

Super-Kamiokande



Function of Photomultiplier



- A photon enters the photocathode, and the photo-electron is emitted by photo-electric effect.
 - Quantum efficiency: 10~30% (typical)
- Inside vacuum
- The first photo-electron is focused into the first dynode.
- Voltage between dynodes is typically 100V or so
- With many dynodes, the number of electrons are multiplicated.
- A typical multiplication factor of one dynode is 2~3, but with 13 dynodes, the multiplication becomes 3¹³~1,000,000.
- It is readout as an electronic signal.

Photomultiplier voltage divider

 High Voltage (HV) divider (Base): Voltage of 1000~2000V is applied. Positive HV type, and the negative HV type exist.



Fig. 8.10a – c. Examples of PM voltage divider networks (after examples from *Philips Catalog* [8.7]: (a) divider network using positive high voltage; note the AC coupling capacitor at the anode, (b) a network using negative high voltage and decoupling capacitors for maintaining the voltages between the last few dynodes,

-HI

The number of photons (Scintillator + PMT)



- Example
 - Incident energy of γ ray: 0.5 MeV
 - The number of scintillation lights: 20,000
 - Quantum efficiency (+light collection in a crystal): 15%
 - Energy resoultion
 - $\sqrt{(3000) \div 3000} = 0.018 (1.8\%)$

Semiconductor photon detector

Hamamatsu HP: "What is MPPC?"

https://www.hamamatsu.com/jp/ja/product/optical-sensors/mppc/what_is_mppc/index.html

	PD	APD	MPPC	PMT
	Photo-diode	Avalanche photo- diode	Multi-channel Giger mode photo-diode	Photomultiplire tube
gain]	10 ²	~106	~107
sensitivity	low	medium	high	high
Voltage applied	5V	100~500 V	30~60 V	800~1000V
Sensitive area	small	small	small-medium	large
Electronics	complicated	complicated	simple	simple
Noise	low	medium	medium	low
Uniformity	\bigcirc	\bigcirc	\bigcirc	\bigcirc
Fast response	\bigcirc	\bigcirc	\bigcirc	\bigcirc
Energy resolution	good	not bad	good	good
Temperature dep.	low	high	medium	high
Outer-light dep.	\bigcirc	\bigcirc	\bigcirc	×
Magneticfield dep.	\bigcirc	\bigcirc	\bigcirc	×
Compact & light	\bigcirc	\bigcirc	\bigcirc	×

Semiconductor photon detector



HAMAMATSU PHOTON IS OUR BUSINESS

ホーム > 製品情報 > 光センサ > 📄

MPPC (SiPM)

Japanese (many master thesis, but one English) https://www-he.scphys.kyoto-u.ac.jp/theses

2008	大谷 将士	T2K長基線ニュートリノ振動実験ニュートリノビームモニター INGRIDの製作と性能評価	2009/01/29	
2000	河崎 直樹	K ^O TO実験のためのNeutron Collar Counterのdesignおよび開発	2000/01/20	
	永井 直樹	T2K実験において用いられる半導体検出器MPPCの大量測定		
	増田 孝彦	K ^O TO実験に用いる低消費電力型光電子増倍管ベースの開発		
	川向 裕之	T2K長基線ニュートリノ振動実験ニュートリノビームモニター INGRIDに用いるシンチレーター及び光子検出器MPPCの性能評価		
	久保 一	NUMIビームラインを用いたT2K実験ミューオンモニターの長期試験 (FNAL T968実験)		
2007	黒澤 陽一	長基線ニュートリノ振動実験T2Kの電磁ホーンの 調整位置/電流モニタ・制御系の開発研究	2008/01/31	
	五味 慎一	半導体光検出器MPPCの性能評価システムの構築		
	塩見 公志	E14実験におけるバックグラウンド事象についての研究		
2006 *	田口 誠	Development of Multi-Pixel Photon Counters and readout electronics	2007/01/31	
	松岡 広大	T2K長基線ニュートリノ振動実験ミューオンモニターの開発	2007/01/31	
	江澤 孝介	ガンマ線に対して高い位置分解能を持ったシンチレーション検出器の開発		
	栗本 佳典	T2K実験におけるニュートリノビームモニターの開発		
2005	中島 康博	中性K中間子稀崩壊探索実験のためのエアロジェルを用いた光子検出器の開発	2006/02/01	
	信原 岳	新型光検出器MPPCの開発		
			98	





Semiconductor detector

- What is semiconductor?
 - Semiconductors have the intermediate remittances between metals and insulators. They are crystalline materials with covalent bonding; silicon and germanium.
- Very good energy resolution
- Doped semiconductors

(1) n-type



Fig. 10.1. Energy band structure of conductors, insulators and semiconductors

- The impurity atom (P, As, Sb, ..) has five valence electrons, an extra electron is left.
- (2) p-type
 - The impurity atom (AI, B, Ga, In, …) has three valence electrons (one less valence electrons), there is an excess of holes.

The number of electrons in the conduction band





p-type



Fig. 10.4. (a) Addition of donor impurities to form n-type semiconductor materials. The impurities add *excess* electrons to the crystal and create donor impurity levels in the energy gap. (b) Addition of acceptor impurities to create p-type material. Acceptor impurities create an *excess* of holes and impurity levels close to the valence band

The np Semiconductor Junction



Fig. 10.5. (a) Schematic diagram of an np junction, (b) diagram of *electron* energy levels showing creation of a contact potential V_0 , (c) charge density, (d) electric field intensity



Semiconductor (The np junction and Depletion depth w/ reversed bias)

<u>http://www.x-ray.co.jp/mame.html</u>



Ge γ ray semiconductor detector





Germanium semiconductor detector



Energy (keV)

Semiconductor detectors

- Semiconductor detectors for vertex and imaging
 - Vertex detector
 - Strip structure
 - Pixel structure
 - X ray CCD





Superconducting detector (Ultimate detector?)



Advantage & Disadvantage of Semiconductor (Superconducting) Detector

- Good points
 - Excellent Energy resolution
 - Precise position resolution
- Weak points
 - Expensive
 - not easy to make a large detector
 - Only experts can produce
 - Experts knowledge is requested even for operation
 - A superconducting detector is operated only in very low temperature.
 - The data size is often huge because of fine segmentation.

Measurements of particles -other type of detectors-

6. Measurements of particles

- 1. Cherenkov Detector
- 2. Transition Radiation Detector
- 3. TPC: Time Projection Chamber
- 4. Nuclear Emulsion
- 5. Nobel liquid detector

Cherenkov light



Kyoto U. Nuclear Reactor



Inside of reactor taken by TN in March, 2014

• When we look into the inside of nuclear reactor, we can see blue light (Cherenkov light).

Cherenkow light detected by Super-Kamiokande

















Cherenkov light

- Cherenkov radiation arises when a charged particle in a material medium moves faster than the speed of light in the same medium.
 - The condition is $v = \beta c > c/n = \beta n > 1$
 - Is is discovered by Cherenkov (the Soviet Union) in 1937.

$$\cos\theta_{\rm c} = \frac{ct/n}{\beta ct} = \frac{1}{\beta n}$$



Fig. 2.9. Cherenkov radiation: an electromagnetic shock wave is formed when the particle travels faster than the speed of light in the same medium

Characteristic of Cherenkov radiation

1. There is a threshold on energy (E_{th}) for Cherenkov radiation

$$E_{\rm th} = m_0 c^2 \left\{ -1 + \sqrt{1 + \frac{1}{n^2 - 1}} \right\}$$

- 2. Very fast (Prompt radiation)
- 3. (Disadvantage) The light yield is small. The fraction of energy transforming to the radiation is only 0.1% or so. (Ref. It is ~10% for scintillation)
 - [Q] How many Cherenkov photons are detected in Super-Kamiokande for an 10 MeV electron.
- 4. The Cherenkov photons are emitted in the direction of the charged particle moving with the angle $\theta_{\rm C}$ to form a cone shape. (Ref. The scintillation lights are emitted isotropic).
- 5. The Cherenkov yield is proportional to $1/\lambda^2$ (λ : Wavelength of Cherenkov light).
- With the above characters, a Cherenkov detector is often used for particle identification with the momentum threshold. It is also unique that the moving direction is determined.

How to find Neutrino



Gig It is The

Material medium for Cherenkov

	n-1	β	
		threshold	
Diamond	1.42	0.41	
glass	0.46~0.75	0.57~0.68	
scintillator	0.58	0.63	
Water	0.33	0.75	
Silica aerogel	0.025~0.075	0.93~0.976	
CO ₂ gas	4.3×10-4	0.9996	
He gas	3.3×10 ⁻⁵	0.99997	

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Belle II Aerogel Ring Image Cherenkov Counter (ARICH)





Transition Radiation

- If a high energy charged particle passes through inhomogeneous media, such as a boundary between two different media (with different dielectric properties), the radiation called "Transition Radiation" is emitted.
 - The radiation of X ray (visible light) is emitted to the very forward (backward) direction of the charged particle.
 - The transition radiation photons are emitted with the angle θ=1/γ to form a cone shape to the direction of the charged particle moving.



RADIOISOTOPES,47,383(1998)

TRD: Transition Radiation Detector

- Total energy of Transition Radiation W is proportional to γ .
 - It is used to identify the particle species with $\beta \sim 1$ (>GeV energy region)
 - X ray is emitted by an electron with >1GeV





$$W = \frac{1}{3} \alpha \hbar \omega_{\rm p} \gamma$$
$$\gamma = \frac{1}{\sqrt{1 - \beta^2}} = \frac{E}{mc^2} :$$
$$\omega_{\rm p} = \sqrt{\frac{N_{\rm e}e^2}{\varepsilon_0 m_{\rm e}}} : \quad \text{Plasma}$$
frequency

https://youtu.be/GOJsccquXNQ



AMS HP

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TPC: Time Projection Chamber

- A three-dimensional tracking detector capable of providing information on many points of a particle track along with information on the specific energy loss, dE/dx, of the particle.
- The TPC makes use of ideas from both the MWPC and drift chamber. ^w
 - Two dimensions by hit positions
 - One dimension by drift time of electrons.



TPC : T2K ND280





TPC : AXEL detector

https://www-he.scphys.kyoto-u.ac.jp/research/Neutrino/AXEL/index.html



Nuclear Emulsion (a kind of Photo film)

- A simple equipment to measure the radiation. It has a long history.
 - In 1896, Becquerel discovered radioactivity by observing blackened Photo film
- It is still the high-end equipment by which the most precise position and the most precise tracking are realized. (Week point: no timing information).
- Once a AgBr crystal absorb a photon, AgBr is decomposed and the region with Ag becomes black.





Nuclear Emulsion

- Since the AgBr crystal particle is about 200nm size, the precision of track position better than 1μ m is possible. (Today, the most precise tracking device).
- No electric power is necessary, and we can just put the films where we like.
 - After the exposure, the development of the films is necessary.
- The image on the film can be scanned and recorded as digital data (image).
- No timing information (integrate all activities [tracks])
 - [Hybrid system] The emulsion is sometimes used together with other detector of good timing and moderate tracking (position) capability.



Crystal of AgBr



Cosmic ray interaction recorded in the emulsion film

Negoya Univ. F-Lab

Nuclear Emulsion for a neutrino experiment



by A. Hiramoto (Kyoto University)

Application of Emulsion

- Cosmic Ray Muon Radiography
 - Inside a volcano, Pyramid, etc..
- Precision Cosmic ray experiment





GRAINE Gamma-Ray Astro-Imager with Nuclear Emulsion

Nagoya Univ. F-Lab
http://www.aip.nagoya-u.ac.jp/en/public/nu_research/highlights/detail/0004155.html

Highlights

Physicists at Nagoya University discover a huge void in Giza's Great Pyramid by cosmic-ray imaging

Read in Japanese

📫 いいね!

🎔 ツイート

2017/11/22

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Institute of Materials and Systems for Sustainability / Institute of Advanced Research

Designated Associate Professor Kunihiro Morishima



Figure 1. Khufu Pyramid at Giza © Kunihiro Morishima







Nuclear emulsion film

Detectors installed in the Queen's Chamber





a : King's Chamber, b : Grand Gallery



Emulsion as popular trend device



Nobel Liquid detector

- Instead of gas phase, liquid phase of the material is used for a radiation measurement
- Advantage : The density is 1000 times higher.
 - The high interaction probability for rare events (neutrinos, dark matter, etc..)
 - More electrons per unit length (1000 times more than gas). No amplification may be necessary.
- Disadvantage
 - No avalanche (no amplification) occurs in liquid.
 - The density of impurity may be 1000 times more. The long drift length of electrons with lower impurity is the key to operate this kind of detector.

	Z	$\left(\frac{g}{cm^3}\right)$	Boiling point	W(eV/ion&e pair)	
				expect	measure
Lq. Ar	18	1.41	87 K	23.3	23.6
Lq. Kr	36	2.15	120 K	19.5	
Lq. Xe	54	3.52	166 K	15.4	15.6
·			融点		
Solid Ar	18	1.62	84 K		

Liquid Ar TPC



Even neutrinos are seen?

The Argon TPC: This detector has t particles pass the

the track signals when charged argon volume. TPC is

can







by T. Maruyama